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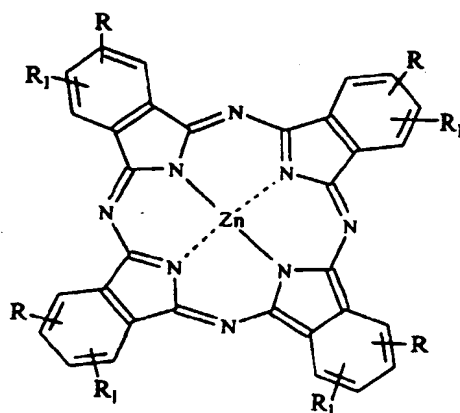
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(54) Zinc-phthalocyanines and corresponding conjugates, their preparation and use in photodynamic therapy and as diagnostic agents

(57) The present invention refers to zinc-phthalocyanines of general formula (I)



(I)

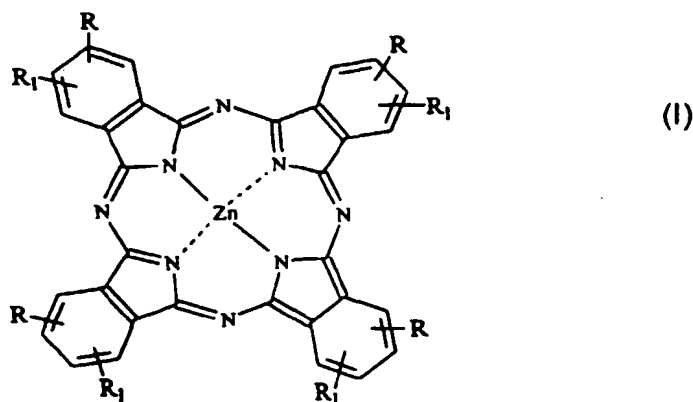
to processes for their preparation and to their use as phototherapeutic and photodiagnostic agents as free molecules and as conjugates with macromolecular carrier molecules

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Description

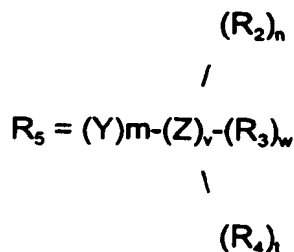
Scope of invention

[0001] The present invention refers to zinc-phthalocyanines of general formula (I)



in which:

R and R₁, same or different from one another, represent H or the group X-R₅, where:
X is chosen in the group consisting of -CH₂-, O, N, S, C=O, and



where:

Y is chosen in the group consisting of C₁₋₁₀-alkyl, phenyl possibly substituted, (CH₂CH₂O)_p, where p ranges from 1 to 4;

Z is chosen in the group consisting of H, N, O, S, SO₃, -CH-, -CH₂- carbon atom, CH₂O, CONH, (CH₂)_qCO₂, where q ranges from 0 to 2;

v is an integer comprised between 1 and 5

R₂ is chosen in the group consisting of H, C₁₋₆-alkyl, PO(OEt)₂, CH₂CH₂NH₂, aryl, and crown ether, or it forms, with the Z group to which it is bound, a saturated or unsaturated heterocycle, possibly substituted, which may contain up to two hetero-atoms chosen in the group consisting of: N, O, and S;

R₃ and R₄, which may be the same or different from one another, are chosen in the group consisting of H, CH₃, and C₂H₅;

m, n, w, t (independently from one another) are 0 or 1;

with the proviso that at least one of R and R₁ is always other than H.

[0002] These products show high photodynamic properties and a marked absorption in the red region of the visible spectrum. These compounds are useful both as such as in the form of conjugates with macromolecular carriers, in the treatment of infectious diseases and diseases characterized by cellular hyperproliferation, and in particular tumours, psoriasis, atheromas, intimal hyperplasia, benign prostate hyperplasia, or else for diagnostic purposes.

State of the art

[0003] It is known that, once organic molecules containing the chromofluorophore macrocycle of the phthalocyanine are photo-activated by irradiation with visible light, they are capable of generating hyper-reactive derivatives of oxygen, above all singlet-oxygen or radicals, which are characterized by a high degree of cytotoxicity, and hence are potentially interesting for therapeutic applications, such as photodynamic therapy (PDT), and/or diagnostic applications (E. Ben-Hur and I. Rosenthal, *Int. J. Radiat. Biol.*, 47 145-147 (1985)).

[0004] The therapeutic application of photosensitizing molecules, prevalently studied in connection with their anti-cancer activity, is based upon the use of photosensitizing agents of a porphyrin nature (Gomer C.J., *Seminars in Hematology*, Vol. 26, pp. 27-34, 1989), which, albeit giving promising results in the curative or palliative treatment of various neoplasms, are markedly limited by the low photosensitization efficiency, poor selectivity, and prolonged persistence in the skin, which may often cause phenomena of generalized photosensitivity (Jori G., *J. Photochem. Photobiol., B: Biol.*, Vol. 36, pp. 87-93, 1996).

[0005] For photosensitizing agents to be usefully employed for therapeutic and/or diagnostic purposes *in vivo*, the following properties are indispensable:

- i) low dark toxicity with a high production of hyperactive derivatives of oxygen, above all singlet-oxygen or radicals, or have a high fluorescence quantum yield;
- ii) selective accumulation by the cells that are responsible for a given pathological condition and fast elimination from the tissues that are not affected by the said pathological condition;
- iii) capacity of being activated by radiation of high wavelength that is able to penetrate more deeply into the tissues as compared with radiation of shorter wavelength;
- iv) possibility of being conjugated to macromolecular carriers, albeit maintaining the characteristics of photosensitization efficiency.

[0006] Thus it is evident how important it is, for therapeutic and/or diagnostic purposes, to be able to develop compounds which, while maintaining the necessary specific biocidal properties, have the characteristics specified above.

[0007] Notwithstanding the advantages that may be foreseen for molecules belonging to the class of the phthalocyanines and possessing their basic chemical structure, only a few of these molecules have been actually assessed as potential agents for photodynamic therapy both on cell lines and *in vivo* at a preclinical level.

[0008] Derivatives that may fall within structure (I) presenting hydroxyl, amine or quaternary ammonium substituents have in fact been described for photosensitization of cancer cells by Leznoff C.C. *et al.* in "Synthesis and photocytotoxicity of some new substituted phthalocyanines", *Photochemistry and Photobiology*, Vol. 49, No. 3, pp. 279-284 (1989), by Wohrle D. *et al.* in "Synthesis of positively charged phthalocyanines and their activity in the photodynamic therapy of cancer cells", *Photochemistry and Photobiology*, Vol. 51, No. 3, pp. 351-356 (1990), again by Wohrle D. *et al.* in "Efficient synthesis of phthalocyanines and related macrocyclic compounds in the presence of organic base", *Dyes and Pigments*, Vol. 18, pp. 91-102 (1992), and by Dumin H. in "Selective photosensitization of mitochondria in HeLa cells by cationic Zn(II) phthalocyanines with lipophilic side-chains", *J. Photochem. Photobiol.*, 37(3) 219-229 (1997).

[0009] Experiments of cancer phototherapy with phthalocyanines on laboratory animals have been reported by Barr H. *et al.* in "Photodynamic Therapy for Colorectal Cancer: a quantitative pilot study", *Br. J. Surg.*, Vol. 77, pp. 93-96, 1990, by Schieweck K. *et al.* in "Liposome-delivered Zn-phthalocyanine as a phototherapeutic agent for tumours", *Proc. SPIE*, Vol. 2078, pp. 107-118, 1994, Ometto C. *et al.* in "Tumour-localizing and photosensitizing properties of a novel phthalocyanine", *Br. J. Cancer*, Vol. 74, pp. 1891-1899, 1996, and by Rousseau J. *et al.* in "Synthesis, tumour uptake and biodistribution of C14-labeled di- and tri-sulfonated gallium phthalocyanine", *J. Photochem. Photobiol., B: Biol.*, Vol. 6, pp. 121-132, 1990.

[0010] Minnoch *et al.* (*J. Photochem. and Photobiol.*, Vol. 32, No. 3, pp. 159-164, 1996) and Brown S. *et al.* (*Photochem. and Photobiol.*, 65(3) pp., 1967) have described the *in vitro* activity of four phthalocyanine derivatives both on micro-organisms and on cell lines. However, the quaternary ammonium compound which is reported as being the only one active in the series of molecules synthesized and assessed against Gram-negative and Gram-positive bacteria, consists of a mixture of compounds having different number of substituents on the phthalocyanine-moiety, and hence is not a compound defined by a precise structural identity (Griffith J. *et al.*, *Dyes & Pigments*, 33(1) 65 (1997)) unlike the products described in the present invention.

Detailed description of the invention

[0011] The present invention makes it possible to meet the above mentioned requirements by means of zinc-phthalocyanines of general formula (I) as defined previously.

[0012] These compounds present a considerable photosensitizing activity which enables them to be used in PDT of

tumoral forms and other illnesses characterized by cellular hyperproliferation, but also against viral, fungal or bacterial diseases; moreover, in so far as they are fluorophores, they may be used as diagnostic agents for the identification of areas that are pathologically affected.

[0013] The presence of the substituents indicated, preferably of a hydrophilic nature, and/or the conjugation to hydrophilic carriers, can, among other things, accelerate the metabolism of the molecule, enabling a fast *in vivo* elimination of the chromophore, and thus preventing the onset of cutaneous phototoxicity.

[0014] According to the invention, by C₁₋₁₀-alkyl group the following are meant: methyl, ethyl, propyl, isopropyl, butyl, pentyl, hexyl, heptyl, octyl, nonyl, and decyl.

[0015] By saturated or unsaturated heterocycle possibly substituted, as defined in the general formula, the following are preferably meant: morpholine, piperidine, pyridine, pyrimidine, piperazine, pyrrolidine, and pyrroline.

[0016] According to the invention, the preferred products are those in which the group X-R₅ is represented by:

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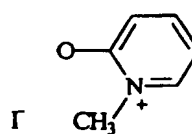
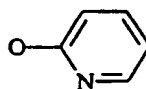
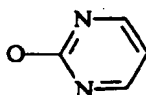
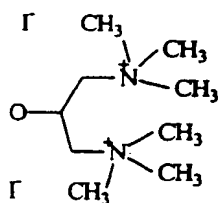
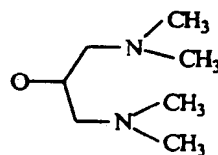
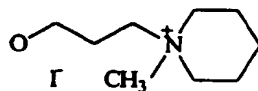
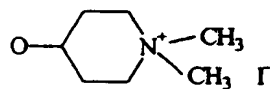
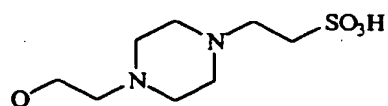
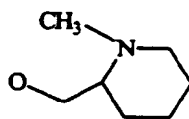
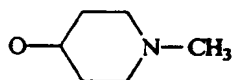
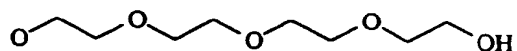
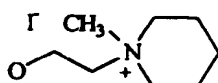
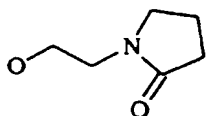
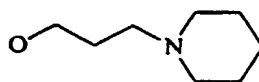
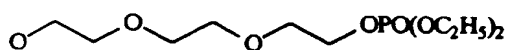
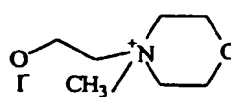
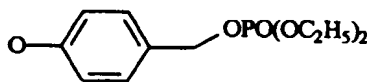
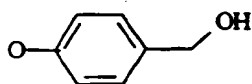
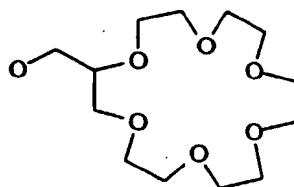
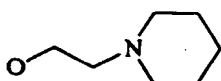
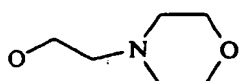
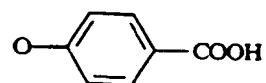
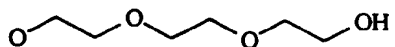
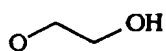
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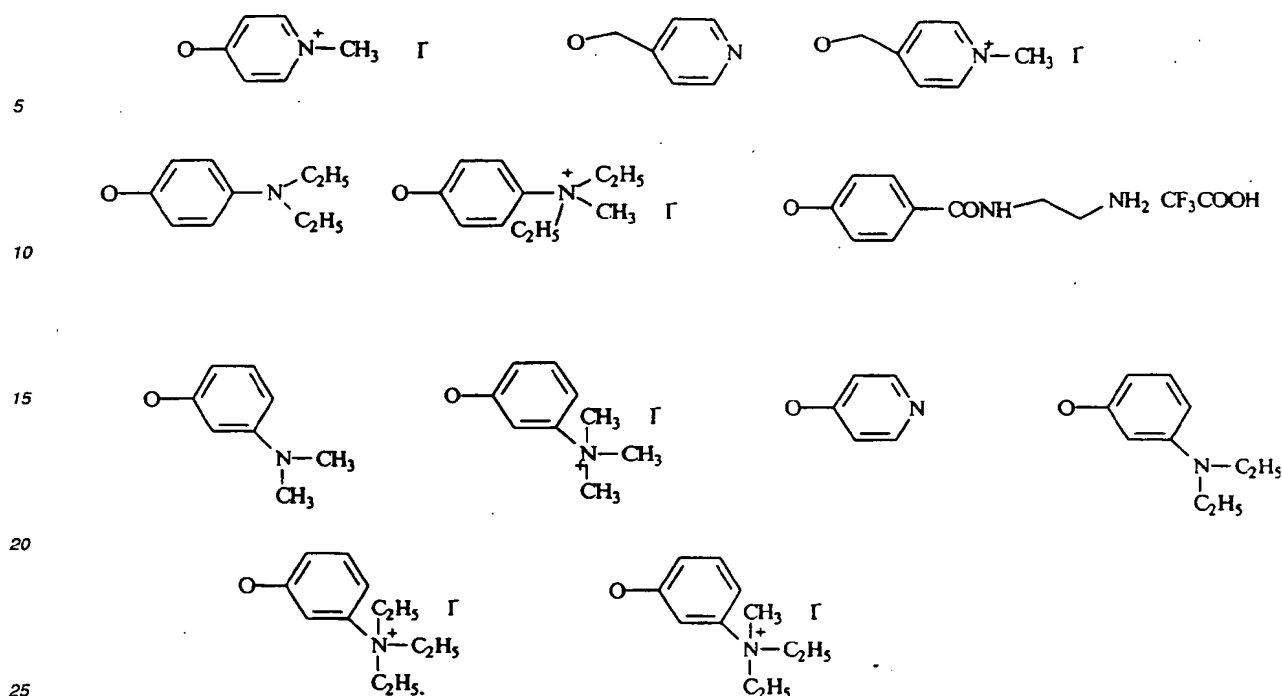
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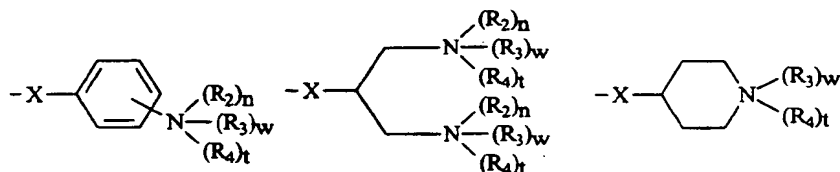
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[0017] In particular, the products in which the group X-R₅, as specified above, contains substituents bearing tertiary or quaternary nitrogen are preferred. Still more preferred are the compounds of formula I wherein the group X-R₅ is a group of formula



[0018] The new synthesized products have high molar extinction coefficients, a characteristic that enables the possibility of an effective therapeutic response at low dosages of active compound and show an unusually high values of the singlet oxygen quantum yields.

[0019] These products may be activated by tissue penetrating radiation having a wavelength of over 650 nm, and hence are suitable for the PDT of various diseases, both dermatological and internal.

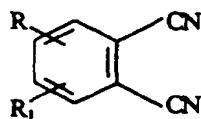
[0020] Normally the present compounds are hydrophobic, if not they can be easily transformed into hydrophilic derivatives by conjugation with hydrophilic macromolecular carriers and are not aggregate in water solution. The products formed by photobleaching of those compounds are non toxic. This finding reinforces their usefulness as therapeutics since after having expleted their action the compounds are inactivated by the light and then no more potentially toxic *in vivo*.

[0021] It must be remembered that the photodynamic activity in some cases and for same substituents is exerted even at low oxygen concentration, this finding enable the use of such derivatives for the specific treatment of anaerobic microorganisms or the treatment of tumor diseases known to be characterized by an hypoxic environment.

[0022] The activity as photosensitizers, the wavelength absorption shifted towards the red and their fluorescence, makes these molecules particularly interesting for use in photodynamic therapy and as tracers for *in vivo* diagnostics.

[0023] The compounds of the present invention can be prepared, in the homogenous phase, according to reaction schemes that are known in organic chemistry.

[0024] The preparation of phthalocyanines was carried out via tetramerization (see Scheme 4) of the corresponding phthalonitrile of formula (II)



(II)

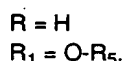
where the substituents R and R₁ are as defined above, using procedures that are known in the literature (Phthalocyanines - Properties and Applications, Vol. 1-4, C.C. Leznoff and A.B.P. Lever VCH Ed.).

[0025] Described in the literature are reactions of synthesis of phthalocyanines having one substituent different from the other three, starting from a dinitrile bound to the solid phase with a dinitrile differently substituted, after prior transformation into diaminoisoidolyl derivatives [Tetrahedron Letters, 23(30) 3023-3026 (1982); J. Org. Chem., 56, 82-90 (1991)].

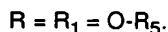
[0026] We have surprisingly found that, following the procedure described in Example 7, also phthalocyanines of formula (I) having four identical substituents may be obtained in heterogeneous phase using polystyrene-based matrices. The condensation between one another of the dinitriles bound to the resin takes place also in the presence of a considerable excess of the dinitrile, variously substituted, in the solution, unlike what is reported in the literature.

[0027] Schemes 1-3 illustrate, as non-limiting examples, three different procedures that make it possible to obtain 3-, 4-, 4,5 and 3,6-substituted phthalonitriles, respectively.

[0028] Scheme 1 refers to the particular case in which:



[0029] Schemes 2 and 3 refers to the case in which:



[0030] As may be seen in the said schemes, the 3- and 4-substituted phthalonitriles were prepared starting from nitrophthalonitriles which were reacted with an appropriate alcohol in DMSO, in the presence of potassium carbonate, for 2 - 72 hours.

[0031] The 4,5- disubstituted phthalonitriles are obtained starting from 4,5- dichloro phthalonitrile in a similar way.

[0032] The 3,6-disubstituted phthalonitriles are obtained by reaction of the corresponding 2,3-dicyanohydroquinones with an appropriate chloro-, bromo-, tosyl-, or mesyl-derivative (R₅Q, where Q is Cl, Br, tosyl or mesyl), in the presence of a base, solvents, and appropriate temperatures according to the particular case.

[0033] The present invention also comprises products of formula (I) as defined above, conjugated to a macromolecular carrier to improve the pharmacological characteristics of the latter.

[0034] The carrier is normally chosen in the group consisting of amino acids, polypeptides, proteins, and polysaccharides.

[0035] The phthalocyanine (I)/carrier bond may occur between the corresponding carboxyl or amine groups or by exploiting other known functional and reactive groups, whether homo- or hetero-bifunctional.

[0036] In order to provide a better illustration of the invention, a number of specific examples are given of the synthesis of products of formula (I) and formula (II).

[0037] In the formulas that follow, the definitions of a1, a2, a3 and a4 are as indicated below:

- a1: Zn-phthalocyanine tetrasubstituted in 2(3), 9(10), 16(17), 23(24);
- a2: Zn-phthalocyanine tetrasubstituted in 1(4), 8(11), 15(18), 22(25);
- a3: Zn-phthalocyanine octasubstituted in 1, 4, 8, 11, 15, 18, 22, 25;
- a4: Zn-phthalocyanine octasubstituted in 2, 3, 9, 10, 16, 17, 23, 24.

EXAMPLE 1

Synthesis of the compound (II) in which $OR_5 = 1\text{-methylpiperidinyl-4-oxy}$ (see Scheme 1)

[0038] 0.173 g of 3-nitrophthalonitrile (1 mmol) are solubilized in 2 ml of DMSO; to the solution are added 0.173 g of 4-hydroxy-*N*-methyl piperidine (1.5 mmol) and 1.24 g of K_2CO_3 . The product is reacted under stirring at room temperature for 72 hours. The solution is filtered, the solvent removed, and the crude reaction product is purified by column chromatography, eluting with a mixture of ethyl acetate (6)/methanol (1)/triethylamine (10), yield, 50%; MW: 241.31.

EXAMPLE 2

Synthesis of the compound (I) a2 in which $R = 1\text{-methylpiperidinyl-4-oxy}$; $R_1 = H$ (Compound 1)

[0039] 0.060 g of the compound (II), prepared according to Example 1 (0.25 mmol), are solubilized in 3 mL of *n*-pentanol in presence of lithium in excess. The mixture is heated while stirring under inert gas up to 150°C for 15 minutes. The resulting lithium phthalocyanine is transformed into the corresponding metal-free compound by treatment with acids (AcOH: pH of 4-5). Then the green solid is stirred with $Zn(OAc)_2$ in DMF at 80°C for 20 hours. Formula: $C_{56}H_{60}N_{12}O_4Zn$; light blue solid; MW, 1030.56; UV-vis λ_{max} (DMF) 701, 632, 380 nm.

EXAMPLE 3

Synthesis of the compound (II) in which $OR_5 = 2\text{-(piperidin-1-yl)ethoxy}$ (see Scheme 1)

[0040] The procedure according to Example 1, starts from 0.089 g of 4-nitrophthalonitrile (0.51 mmol), 0.1 g of *N*-(2-hydroxyethyl)-piperidine (0.77 mmol) and 0.515 g of K_2CO_3 (3.69 mmol). The reaction is stopped after 24 hours by adding water and extracting the product with CH_2Cl_2 . The organic extracts are dried on Na_2SO_4 , evaporated to obtain a solid, which is purified by column chromatography (Silica gel; ethyl acetate 4/methanol 1). In this way, 0.1 g of yellowish solid product is obtained (yield 78%).

EXAMPLE 4

Synthesis of the compound (I) a1 in which $R = 2\text{-(piperidin-1-yl)ethoxy}$; $R_1 = H$ (Compound 2)

[0041] 0.228 g of the compound (II), prepared according to Example 3 (0.89 mmol), are reacted with $Zn(OAc)_2$ in the presence of 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) under inert gas at 160°C for 4 hours. The crude reaction product is purified by extraction with solvents (water/chloroform distribution) and by chromatography (70% yield). Formula: $C_{60}H_{68}N_{12}O_4Zn$; green solid; UV-vis λ_{max} (THF) (ϵ) 679 (140320), 612, 354 nm; EAS m/z 1085.6 ($M+H$)⁺.

[0042] Following a similar procedure, the following products were obtained:

Product (I) a1 in which $R = 2\text{-(morpholin-4-yl)ethoxy}$; $R_1 = H$ (Compound 3) Formula: $C_{56}H_{60}N_{12}O_8Zn$; blue-green solid; UV-vis λ_{max} (DMF) 680, 612, 354 nm; UV-vis λ_{max} (THF) (ϵ) 675 (159410); EAS m/z 1093.8 ($M+H$)⁺.

Product (I) a3 in which $R = R_1 = 2\text{-[2-(2-hydroxyethoxy)ethoxy]ethoxy}$ (Compound 4)

Formula: $C_{80}H_{112}N_8O_{32}Zn$; green solid; UV-vis λ_{max} (DMF) 728, 656, 324 nm; EAS m/z 1761.0 ($M+H$)⁺.

Product (I) a2 in which $R = 2\text{-(piperidin-1-yl)ethoxy}$; $R_1 = H$ (Compound 5) Formula: $C_{60}H_{68}N_{12}O_4Zn$; green-blue solid; UV-vis λ_{max} (DMF) (ϵ) 760, 699 (80500), 629, 391 nm; EAS m/z 1085.6 ($M+H$)⁺.

Product (I) a1 in which $R = (18\text{-crown-6})\text{methoxy}$; $R_1 = H$ (Compound 6)

Formula: $C_{84}H_{112}N_8O_{28}Zn$; green solid; UV-vis λ_{max} (DMF) (ϵ) 679 (85910), 611, 356 nm; EAS m/z 1746 ($M+H$)⁺.

Product (I) a1 in which $R = 4\text{-(hydroxymethyl)phenoxy}$; $R_1 = H$ (Compound 7)

Formula: $C_{60}H_{40}N_8O_8Zn$; green solid; UV-vis λ_{max} (DMF) (ϵ) 679 (82970), 611, 356 nm; EAS m/z 1064.1 ($M+H$)⁺.

Product (I) a1 in which $R = 4\text{-(diethylphosphorylmethyl)phenoxy}$; $R_1 = H$ (Compound 8)

Formula: $C_{78}H_{76}N_8O_{20}P_4Zn$; green solid; UV-vis λ_{max} (DMF) 678, 610, 357 nm.

Product of formula (I) a3 in which $R = R_1 = 2\text{-[2-(2-diethylphosphorylethoxy)ethoxy]ethoxy}$ (Compound 9)

Formula: $C_{112}H_{184}N_8O_{56}P_8Zn$; green solid; UV-vis λ_{max} (DMF) 733, 661, 360 nm.

Product (I) a3 in which $R = R_1 = 2\text{-(morpholin-4-yl)ethoxy}$ (Compound 10)

Formula: $C_{80}H_{104}N_{16}O_{16}Zn$; brownish-green solid; UV-vis λ_{max} (DMF) 736, 661, 349 nm.

Product (I) a3 in which $R = R_1 = 3\text{-(piperidin-1-yl)propoxy}$ (Compound 11)

Formula: $C_{96}H_{136}N_{16}O_8Zn$; green solid; UV-vis λ_{max} (DMF) 818, 742, 666, 350 nm.

Product (I) a1 in which $R = 2\text{-(pyrrolidin-2-one-1-yl)ethoxy}$; $R_1 = H$ (Compound 12)

Formula: $C_{56}H_{52}N_{12}O_8Zn$; blue solid; UV-vis λ_{max} (DMF)(ϵ) 680 (104100), 611, 354 nm; EAS m/z 1085.2 ($M+H$)⁺.

Product (I) a1 in which $R = 2\text{-}\{2\text{-}\{2\text{-}\{2\text{-hydroxyethoxy}\}\text{ethoxy}\}\text{ethoxy}\}$; $R_1 = H$ (Compound 13)

Formula: $C_{64}H_{80}N_8O_{20}Zn$; blue-green solid; UV-vis λ_{max} (DMF) (ϵ) 680 (133120), 679, 611, 355 nm.

Product (I) a2 in which $R = 2\text{-}\{morpholin-4\text{-yl}\}\text{ethoxy}$; $R_1 = H$ (Compound 14)

Formula: $C_{56}H_{60}N_{12}O_8Zn$; green solid; UV-vis λ_{max} (DMF) (ϵ) 698 (125910), 628, 322 nm; EAS m/z 1094.1 ($M+H$)⁺.

Product (I) a1 in which $R = 1\text{-methylpiperidinyl-4-oxy}$; $R_1 = H$ (Compound 15)

Formula: $C_{56}H_{60}N_{12}O_4Zn$; blue solid; UV-vis λ_{max} (DMF) 681, 613, 354 nm.

Product (I) a2 in which $R = (1\text{-methylpiperidin-2-yl})\text{methoxy}$; $R_1 = H$ (Compound 16) Formula: $C_{60}H_{68}N_{12}O_4Zn$; brownish-green solid; UV-vis λ_{max} (DMF) 715, 642, 318 nm.

Product (I) a2 in which $R = 2\text{-}\{4\text{-}\{1\text{-}\{2\text{-sulfoethyl}\}\}\text{piperazinyl}\}\text{ethoxy}$; $R_1 = H$ (Compound 17) Formula: $C_{64}H_{80}N_{16}O_{16}S_4Zn$; blue-green solid; UV-vis λ_{max} (DMF) 701, 631, 322 nm.

Product (I) a1 in which $R = 2\text{-}\{4\text{-}\{1\text{-}\{2\text{-sulfoethyl}\}\}\text{piperazinyl}\}\text{ethoxy}$; $R_1 = H$ (Compound 18) Formula: $C_{64}H_{80}N_{16}O_{16}S_4Zn$; blue-green solid; UV-vis λ_{max} (DMF) 680, 612, 354 nm.

Product of formula (I) a1 in which $R = 1,3\text{-bis-}\{dimethylamino\}\text{propyl-2-oxy}$; $R_1 = H$ (Compound 19) Formula: $C_{60}H_{80}N_{16}O_4Zn$; blue-green solid.

Product (I) a1 in which $R = \text{pyrimidinyl-2-oxy}$; $R_1 = H$ (Compound 20)

Formula: $C_{48}H_{24}N_{16}O_4Zn$; blue-green solid.

Product (I) a1 in which $R = \text{pyridinyl-2-oxy}$; $R_1 = H$ (Compound 21)

Formula: $C_{52}H_{28}N_{12}O_4Zn$; blue-green solid.

Product (I) a1 in which $R = 3\text{-}\{dimethylamino\}\text{phenoxy}$; $R_1 = H$ (Compound 22)

Formula: $C_{64}H_{52}N_{12}O_4Zn$; blue-green solid.

Product (I) a1 in which $R = \text{pyridinyl-4-oxy}$; $R_1 = H$ (Compound 23)

Formula: $C_{52}N_{28}N_{12}O_4Zn$; blue-green solid.

Product (I) a1 in which $R = (\text{pyridin-4-yl})\text{methoxy}$; $R_1 = H$ (Compound 24)

Formula: $C_{56}H_{36}N_{12}O_4Zn$; blue-green solid.

Product (I) a3 in which $R = R_1 = 4\text{-}\{diethylamino\}\text{phenoxy}$ (Compound 25)

Formula: $C_{112}H_{120}N_{16}O_8Zn$; blue-green solid.

Product (I) a1 in which $R = 3\text{-}\{diethylamino\}\text{phenoxy}$; $R_1 = H$ (Compound 26)

Formula: $C_{72}H_{68}N_{12}O_4Zn$; blue-green solid.

Product (I) a4 in which $R = R_1 = 3\text{-}\{dimethylamino\}\text{phenoxy}$ (Compound 27)

Formula: $C_{96}H_{88}N_{16}O_8Zn$; blue-green solid.

Product (I) a2 in which $R = 3\text{-}\{dimethylamino\}\text{phenoxy}$; $R_1 = H$ (Compound 28)

Formula: $C_{64}H_{52}N_{12}O_4Zn$; blue-green solid.

Product (I) a2 in which $R = 3\text{-}\{diethylamino\}\text{phenoxy}$; $R_1 = H$ (Compound 29)

Formula: $C_{72}H_{68}N_{12}O_4Zn$; blue-green solid.

Product (I) a4 in which $R = R_1 = 3\text{-}\{diethylamino\}\text{phenoxy}$ (Compound 30)

Formula: $C_{112}H_{120}N_{16}O_8Zn$; blue-green solid.

EXAMPLE 5

Synthesis of the compound (II) in which $OR_5 = 2\text{-}\{hydroxy\}\text{ethoxy}$ (see Scheme 3)

[0043] 0.32 g of 2,3-dicyanohydroquinone (2 mmol) are suspended in 5 mL of xylene, and 1.0 g of 2-bromoethanol (8 mmol) and 1.0 g of triethylamine (1.37 mL; 10 mmol) are added to the suspension. The product is heated under stirring at 130°C for 24 hours then the xylene is decanted. The residue is then treated with methanol to obtain 0.24 g of white-grey solid (yield, 70%).

MW, 203.19; 1H NMR (DMSO- d_6) δ 7.65 (s, 2H), 4.95 (t, 2H), 4.20 (t, 4H), 3.80 (m, 4H).

EXAMPLE 6

Synthesis of the compound (I) a3 in which $R = R_1 = 2\text{-}\{hydroxy\}\text{ethoxy}$ (Compound 31)

[0044] 0.1 g of the compound (II), prepared according to Example 5 (0.5 mmol), is solubilized in N,N-dimethylethanolamine, ammonia is passed through this solution for 15 minutes at room temperature then the temperature is raised to 140-160°C for 6 hours, continuing the stream of ammonia. The desired product is obtained by flash-chromatography purification on Silica gel in a 25% yield. Formula: $C_{48}H_{48}N_8O_{16}Zn$; green solid; UV-vis λ_{max} (DMF) 741, 666, 357 nm;

EAS m/z 1056.8 (M+H)⁺

EXAMPLE 7

- 5 Synthesis of the compound (I) a1 in which R = N-(2-aminoethyl)benzamido-4-oxy trifluoro acetate; R₁ = H (Compound 32)

a) Functionalization of the polystyrene-based resin with the phthalodinitrile

- 10 **[0045]** 0.159 g (0.078 mmol) of diaminoethane-trityl resin (0.49 mmol/g) are swelled in 12.5 mL of DMF. To this suspension 0.282 g (0.78 mmol) of the succinimide ester of the compound (II) in which R₅ = 4-carboxyphenyl are added, and the product is kept under stirring at room temperature for 18 hours. The liquid phase is removed from the resin by vacuum filtration, and the resin is washed several times with small volumes of DMF, CH₂Cl₂ and MeOH.

- 15 b) Solid-phase condensation reaction

- [0046]** The obtained functionalized resin (0.078 mmol) is swelled in 2 mL of DMF for one hour at 50°C. 0.080 g (0.43 mmol) of zinc(II)acetate and 0.322 mL (2.15 mmol) of DBU are added and the suspension is heated up to 160°C for 4 hours, under stirring and nitrogen. After cooling at room temperature, the two phases are separated by vacuum filtration, and the solid phase is washed with MeOH and DMF.
- 20

c) Separation of the Zn-phthalocyanine from the resin

- [0047]** The green-blue resin is suspended in a solution of trifluoroacetic acid (TFA) (5%) and tri-isopropyl silane (TIS) (5%) in CH₂Cl₂ and kept in this solution for 1.5 hours. The two phases are then separated by vacuum filtration and the resin is washed with CH₂Cl₂. The solid phase is treated with small volumes of DMF and MeOH alternatively until the solution is colourless. The product is obtained by concentration of the filtrate in a 60% yield. Formula: C₆₈H₅₆N₁₆O₈Zn (CF₃COOH)₄.
- 25

- 30 UV-vis λ_{max} (DMF) 676, 609, 349 nm.

Using procedures known in the literature, the following products were obtained: Product (I) a1 in which R = 2-(morpholin-4-yl)ethoxy methylammonium iodide; R₁ =

H (Compound 33) Formula: C₆₀H₇₂I₄N₁₂O₈Zn; blue-green solid; UV-vis λ_{max} (DMF) (ε) 677 (167330).

Product (I) a1 in which R = 2-(piperidin-1-yl)ethoxy methylammonium iodide; R₁ =

- 35 H (Compound 34) Formula: C₆₄H₈₀I₄N₁₂O₄Zn; blue solid; UV-vis λ_{max} (DMF)(ε) 678 (144840), 611, 353 nm.

Product (I) a2 in which R = 2-(piperidin-1-yl)ethoxy methylammonium iodide; R₁ =

H (Compound 35) Formula: C₆₄H₈₀I₄N₁₂O₄Zn; green solid; UV-vis λ_{max} (DMF) 753, 701, 731 nm.

Product (I) a2 in which R = 2-(morpholin-4-yl)ethoxy methylammonium iodide; R₁ =

H (Compound 36) Formula: C₆₀H₇₂I₄N₁₂O₈Zn; green solid.

- 40 Product (I) a3 in which R = R₁ = 2-(morpholin-4-yl)ethoxy methylammonium iodide (Compound 37) Formula: C₈₈H₁₂₈I₈N₁₆O₁₆Zn; green solid.

Product (I) a1 in which R = 1-methylpiperidinyl-4-oxy methylammonium iodide; R₁ =

H (Compound 38) Formula: C₆₀H₇₂I₄N₁₂O₄Zn; green solid.

Product (I) a3 in which R = R₁ = 3-(piperidin-1-yl)propoxy methylammonium iodide (Compound 39) Formula: C₁₀₀H₁₅₂I₈N₁₆O₈Zn; green solid.

- 45 Product (I) a1 in which R = 1,3-bis-(dimethylamino)propyl-2-oxy dimethylammonium iodide; R₁ = H (Compound 40) Formula: C₆₈H₁₀₄I₈N₁₆O₄Zn; blue-green solid.

Product (I) a1 in which R = piridinyl-2-oxy methylammonium iodide; R₁ = H (Compound 41) Formula: C₅₆H₄₀I₄N₁₂O₄Zn; blue-green solid.

- 50 Product (I) a1 in which R = 3-(dimethylamino)phenoxy methylammonium iodide; R₁ = H (Compound 42) Formula: C₆₈H₆₄I₄N₁₂O₄Zn; blue-green solid.

Product (I) a1 in which R = piridinyl-4-oxy methylammonium iodide; R₁ = H (Compound 43) Formula: C₅₆H₄₀I₄N₁₂O₄Zn; blue-green solid.

Product (I) a1 in which R = (piridin-4-yl)methoxy methylammonium iodide; R₁ = H (Compound 44) Formula: C₆₀H₄₈I₄N₁₂O₄Zn; blue-green solid.

- 55 Product (I) a3 in which R = R₁ = 4-(diethylamino)phenoxy methylammonium iodide (Compound 45) Formula: C₁₂₀H₁₄₄I₈N₁₆O₈Zn; green solid.

Product (I) a1 in which R = 3-(diethylamino)phenoxy methylammonium iodide; R₁ = H (Compound 46) Formula:

$C_{76}H_{80}I_4N_{12}O_4Zn$; green solid.

Product (I) a4 in which $R = R_1 = 3$ -(dimethylamino)phenoxy methylammonium iodide (Compound 47) Formula: $C_{104}H_{114}I_8N_{16}O_8Zn$; green solid.

Product (I) a4 in which $R = R_1 = 3$ -(diethylamino)phenoxy methylammonium iodide (Compound 48) Formula: $C_{120}H_{144}I_8N_{16}O_8Zn$; green solid.

Product (I) a4 in which $R = R_1 = 3$ -(diethylamino)phenoxy ethylammonium iodide (Compound 49) Formula: $C_{128}H_{160}I_8N_{16}O_8Zn$; green solid.

Product (I) a2 in which $R = 3$ -(diethylamino)phenoxy methylammonium iodide; $R_1 = H$ (Compound 50) Formula: $C_{76}H_{80}I_4N_{12}O_4Zn$; green solid.

Product (I) a1 in which $R = 3$ -(diethylamino)phenoxy ethylammonium iodide; $R_1 = H$ (Compound 51) Formula: $C_{80}H_{88}I_4N_{12}O_4Zn$; green solid.

Product (I) a2 in which $R = 3$ -(diethylamino)phenoxy ethylammonium iodide; $R_1 = H$ (Compound 52) Formula: $C_{80}H_{88}I_4N_{12}O_4Zn$; green solid.

Product (I) a2 in which $R = 3$ -(dimethylamino)phenoxy methylammonium iodide; $R_1 = H$ (Compound 53) Formula: $C_{68}H_{64}I_4N_{12}O_4Zn$; green solid.

EXAMPLE 8

General synthesis of amine or carboxylic derivative of formula (I) bound to polypeptides

[0048] 200 μ L of a 5 mg/ml solution of bovine serum albumin (BSA) in PBS (pH: 8.5) containing 25 equiv. of I added as solution in DMSO is incubated for 10 minutes at room temperature. 25 equiv. of a water soluble carbodiimide are added slowly to the sample refrigerated at 4°C and the reaction mixture is gently stirred for 30 minutes, while the temperature reaches room temperature.

[0049] The conjugation product is purified by gel filtration (Sephadex G25) eluting with PBS pH: 7.2, and collecting the fractions containing the conjugate, which are visible owing to the green-blue colour, and from which the solid product is obtained by liophilization.

[0050] The protein concentration and the number of moles of the Compound (I) introduced per mole of BSA (labelling ratio) may be determined spectrophotometrically and it was found to range from 3.5 to 10 moles of phthalocyanine per mole of BSA.

Pharmaceutical formulations

[0051] Therapeutic compositions containing the compounds of the present invention include liposomes or microcapsules, dispersions, solutions for parenteral injection, preparations for topical application, etc.

[0052] The topical formulations according to the invention are, for example, lotions, creams, ointments or gels.

[0053] Particularly preferred are DMSO or Azone aqueous solutions, up to 50 wt%.

[0054] The compounds of the present invention having lipophilic characteristics may be incorporated in liposomes or microcapsules and used in this form for both types of application mentioned above.

[0055] The photodynamic therapy that uses the compounds of the present invention affords numerous advantages.

[0056] They are not toxic in the absence of light, and hence in the non-excited state.

[0057] Each molecule may be repeatedly excited, with the consequent production of singlet oxygen or other reactive species, which entails lethal effects for the cells.

[0058] Given their short average life time they hit the target cell without possibility of affecting vicinal cells.

[0059] The photodynamic therapy that uses the present compounds is thus selective and non-toxic, in that the singlet oxygen produced that does not react with biological targets undergoes a rapid decay; in fact, the production of oxygen takes place immediately after irradiation and stops as soon as irradiation is interrupted.

[0060] The dosages normally range from 0.1 to 20 mg of compound of formula (I) per kilogram of body weight, preferably 0.2-5 mg/kg of body weight.

[0061] The appropriate light sources required to carry out PDT are well known to the art and may, for instance, be white light associated to suitable filters or laser light having the specific wavelength required, with wavelengths between 600 and 950 nm, preferably 650-750 nm.

[0062] The total applied amount of radiation varies according to the treatment and the location of the tissues to be treated.

[0063] Generally the amount of radiation is between 50-1000 J/cm², preferably between 100 and 350 J/cm².

BIOCIDAL ACTIVITY

[0064] The compounds synthesized have been assayed for their antifungal and antibacterial (Gram-positive and Gram-negative) activity. For the experiments the following microorganisms were used: *Candida albicans* (yeast), *Staphylococcus aureus* (Gram-positive) and *Pseudomonas aeruginosa*, as well as *E.coli*, *Porphyromonas gingivalis*, *Branhamella catarralis* (Gram-negative).

[0065] All the micro-organisms were used in the experiments in a stationary state of growth. The experimental protocol was the following:

[0066] Dilution of the cell suspension in the range $10^6 \pm 10^9$ UFC/mL in the appropriate medium. Addition of an aliquot of stock solution of the compound to be tested to the cell suspension up to the intended final concentrations. Incubation in dark at 37°C (5 min to 1 hour). Irradiation ($625 \leq \lambda \leq 850$ nm; 10 ± 100 mW/cm²; 1-30 min) of cell suspension for each dilution of photosensitizing agent. Taking the exposed samples at the time intervals given in the tables, plating and incubating at 37°C in the appropriate culture medium for the specific microorganism, for the colonies to be counted.

[0067] As an example, photoinactivation of some Gram-positive and negative microorganisms by using some compounds of the present invention are given in TABLE A.

[0068] Diagram 1 shows the variation of colony forming units (CFU) as a function of administered light dose; while Diagram 2 shows the cell survival (in %) as a function of the photosensitizer concentration.

[0069] As an example the usefulness of derivatives (I) in the treatment of cell hyperproliferation and psoriasis is demonstrated in Tables B and C where it is shown the in vivo erythematous effect produced by compounds 15 and 42 and the skin recovery after treatment within 10 days. This finding thus support the possibility to use such compounds for the elimination of epidermal cells responsible for the occurrence of the above mentioned pathologies.

[0070] The pharmacokinetic of compound 15 reported in Table D, still as an example, demonstrates the selective photosensitizer uptake by epidermal cells and neither appreciable systemic absorption nor localization in tissues other than skin. Compounds are almost totally eliminated from the application site within 3 hours.

[0071] Finally, Table E gives the difference in the *E.coli* survival as post irradiation times by using compound 40 in comparison with a previously described phthalocyanine derivative, while Table F accounts for the high affinity of compounds 40 and 42 on *Staphylococcus aureus*, *E.Coli* and *Branhamella catarralis*, evaluated as percentage of bound compound after incubation followed from up to three washing steps.

TABLE A

Photoinactivation of microorganisms <i>Staphylococcus aureus</i> (ATCC 6538P)			
Compound	Concentration (μ M)	mW/cm ² (min.)	Cell mortality
3	0.6	15(30)	99.00
2	0.5	15(30)	99.99
6	0.6	15(30)	99.99
ZnPC	1	15(30)	99.99
40	1	100 (1)	99.999
42	1	100 (1)	99.9998
ZnPC (Commercial Zn-Phthalocyanine)			
<i>Pseudomonas aeruginosa</i> (ATCC 9027)			
Compound	Concentration (μ M)	mW/cm ² (min)	Cell mortality
33	28	18 (30)	37
34	8	18 (30)	21
<i>E.Coli</i> (strain 04)			
Compound	Concentration (μ M)	mW/cm ² (min)	Cell mortality
40	1	100 (1)	99.9999
ZnPC was found not active			
<i>Branhamella catarrhalis</i>			
Compound	Concentration (μ M)	mW/cm ² (min)	Cell mortality
40	1	100 (1)	99.9999
42	1	100 (1)	99.98
40	0.5	100 (1)	99.99
42	0.5	100 (1)	99.75
ZnPC was found not active			

TABLE B

Laser irradiation after topical application of Compound 15 (water/5% DMSO)					
Hours after irradiation	I	II	III	IV	V
24 h	+++ and edema in all area	+++ and edema in all area	+++ and edema in all area	+++ and edema in all area	+++ and edema in all area
48 h	*+++	*+++	*+++	*+++	*+++
73 h	*+++	*+++	*+++	*+++	*+++
4 days	+++	+++	+++	+++	+++
5 days	++	+++	+++	++	+++
7 days	++	++	++	+	++

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TABLE B (continued)

Laser irradiation after topical application of Compound 15 (water/5% DMSO)					
Hours after irradiation	I	II	III	IV	V
24 h	+++ and edema in all area	+++ and edema in all area	+++ and edema in all area	+++ and edema in all area	+++ and edema in all area
10 days	+	+	++	+	+

TABLE C

Laser irradiation after topical application of Compound 42 (water/5% DMSO)					
Hours after irradiation	I	II	III	IV	V
24 h	++	++	++	+++	++
48 h	+++	+++	+++ and edema surrounding all area	+++	+++
73 h	+++	+++	+++ and edema surrounding all area	+++	+++
4 days	+++	+++	+++ and edema surrounding all area	+++	+++
5 days	+++	++	+++	+++	++
6 days	++	+	+++	+++	+
7 days	+/-	+/-	+++	++	+/-
8 days	+	0	++	++	0
10 days	+/-	-	+/-	+	0

I, II, III, IV, V indicate the number of experiments.

Experimental conditions: Fluence rate 150 mW/cm² for 13.3 min (Total energy 120 J)

At the 10th day animals were sacrificed

Legend:

- no response
- 0 weak response (erythema and or edema)
- +/- superficial eschar on part of treated area
- + superficial eschar on all the area
- ++ deep eschar on part of the area
- +++ deep eschar covering all the area
- *+++ very deep eschar

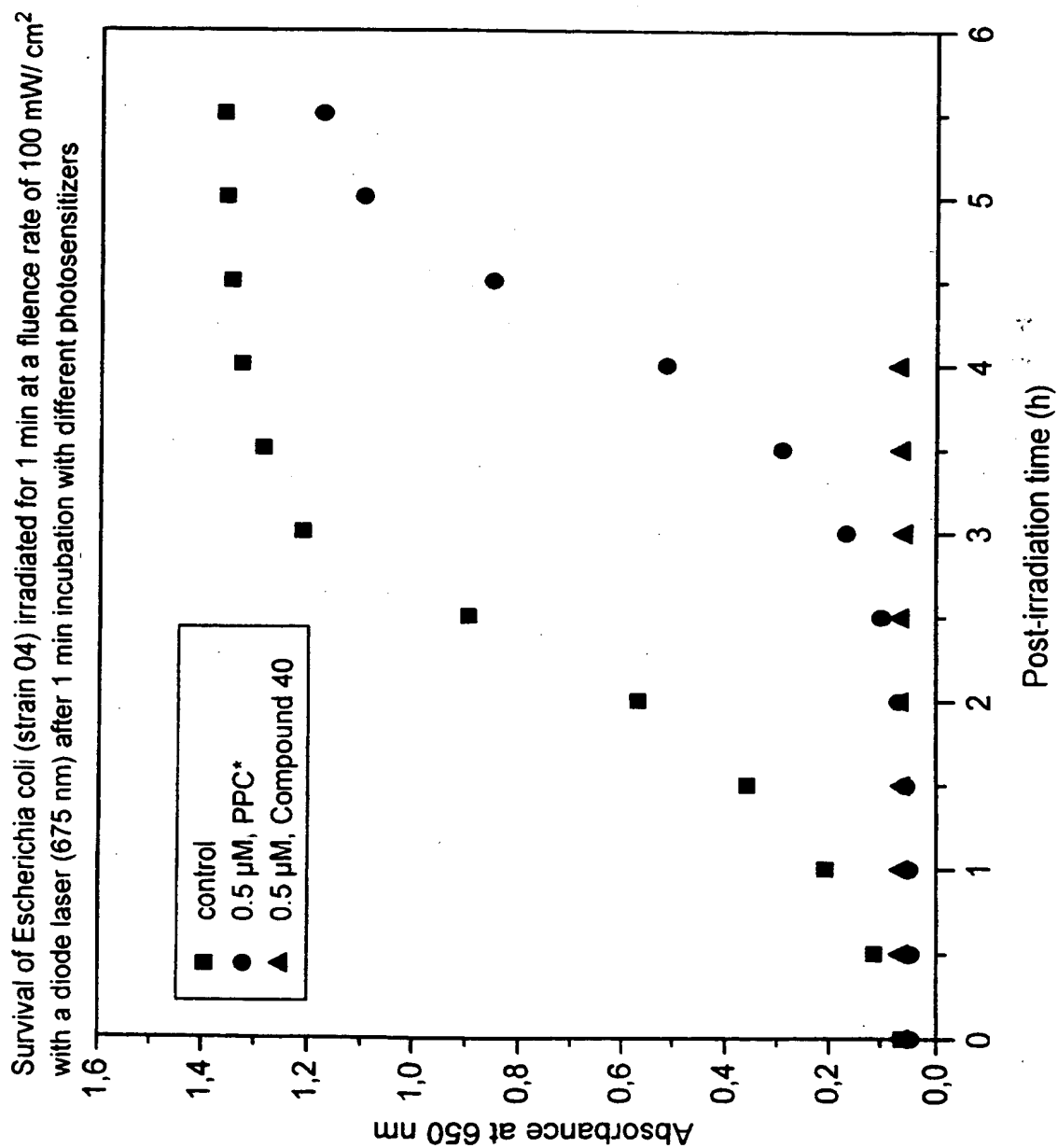
TABLE D

Pharmacokinetics of Compound 15 topically administered (20 µg/cm ² , water/5% DMSO)				
	I	II	III	Average accumulation (%)
Serum 1h	0	0	0	0
Skin 1h	1547.64 ng/cm ²	2278.60 ng/cm ²	2330.00 ng/cm ²	10.23
Liver 1h	0	0	0	0
Serum 3 h	0	0	0	0
Skin 3 h	593 ng/cm ²	690 ng/cm ²	562 ng/cm ²	0.03

TABLE D (continued)

Pharmacokinetics of Compound 15 topically administered (20 µg/cm ² , water/5% DMSO)				
	I	II	III	Average accumulation (%)
Liver 3 h	0	0	0	0
I, II, III indicate the number of experiments.				

TABLE E



*PPC: reported by A. Minnock et al. J. Photochem. Photobiol. 32: 159-164, (1996).

TABLE F

Recovery of Compound 40 and Compound 42 (nmoles/mg of cell protein) bound to <i>Staphylococcus aureus</i> , <i>Escherichia coli</i> , <i>Branhamella catarralis</i> after 5 min. of incubation and washing.				
Microorganism	compound N° (conc. μM)	Recovery of phthalocyanine (nmoles/mg of cell protein) after the following washing steps		
		0	1	3
<i>Staphylococcus aureus</i>	40 (5 μM)	100%	82%	56%
	42 (5 μM)	100%	100%	84%
<i>Escherichia coli</i>	40 (10 μM)	100%	56%	32.4%
	42 (10 μM)	100%	72%	52%
<i>Branhamella catarralis</i>	40 (5 μM)	100%	70%	67%
	42 (5 μM)	100%	92%	90%

DIAGRAM 1

Survival curves of *Candida albicans* incubated with Compound 42 (1µM)

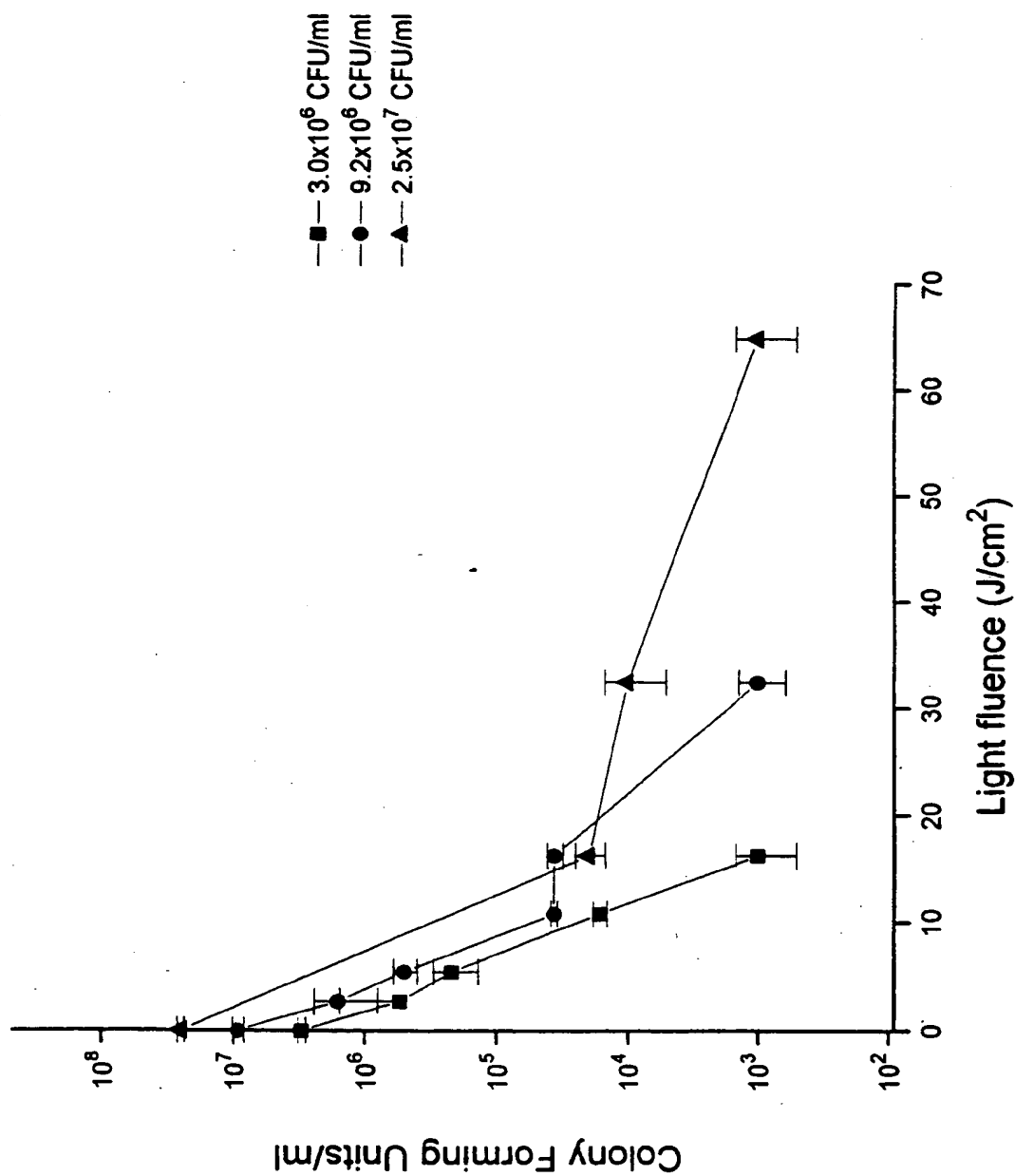
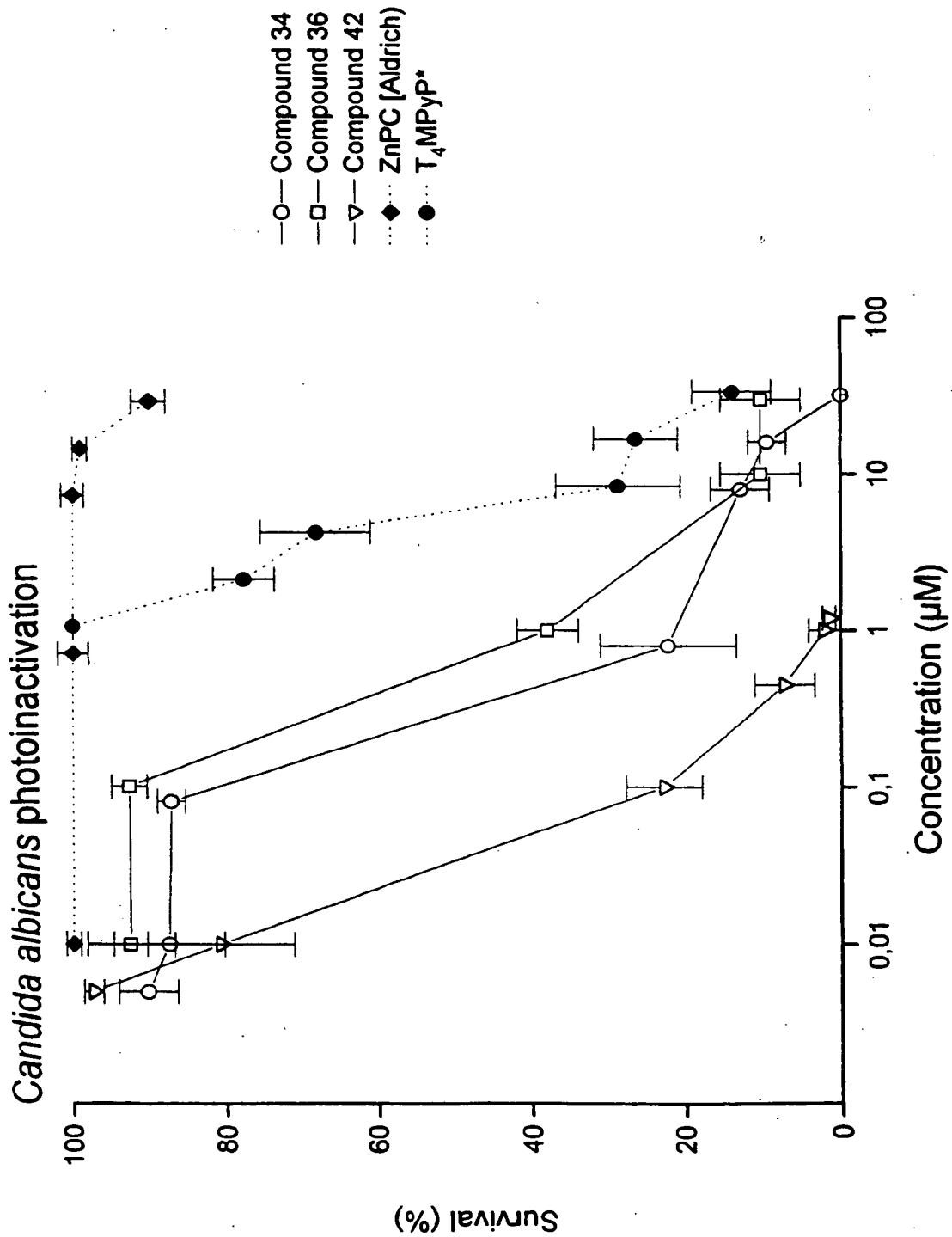
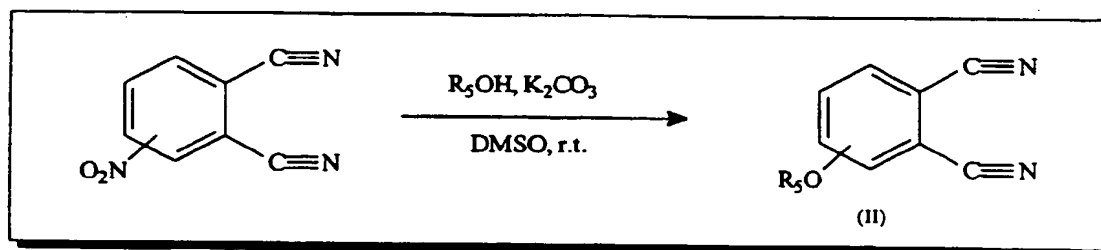


DIAGRAM 2

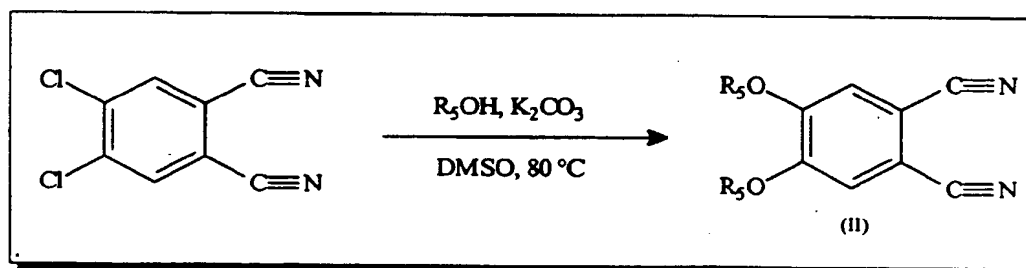


* T_4MPyP^* : reported by M. Merchat et al. J. Photochem. Photobiol. 32: 153-157, (1996).

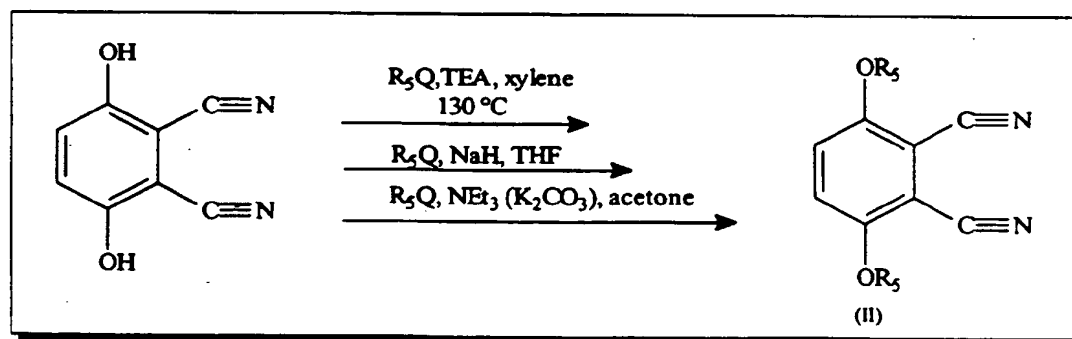
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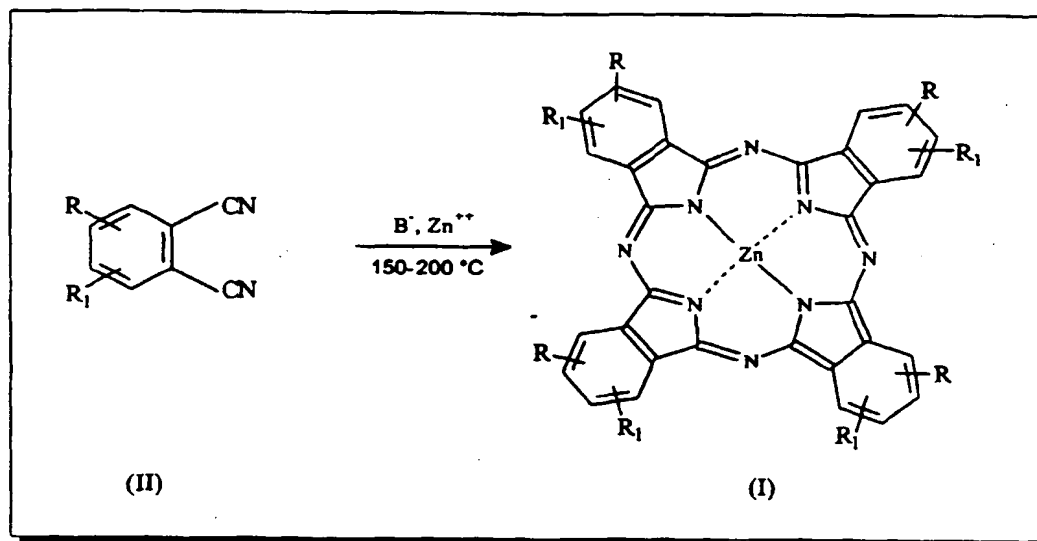
SCHEME 2



SCHEME 3

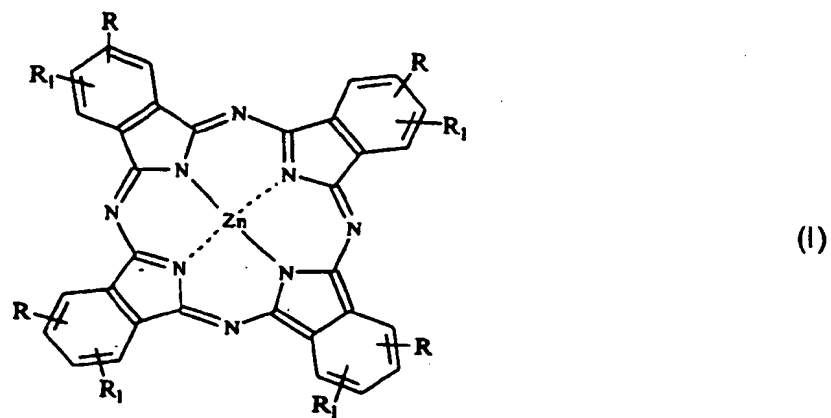


SCHEME 4



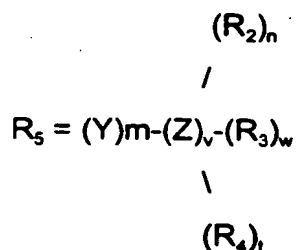
Claims

1. Compounds of general formula (I)



in which:

R and R_1 represent H or the group $X-R_5$, where:
 X is chosen in the group consisting of $-CH_2-$, O, N, S, $C=O$, and



where:

Y is chosen in the group consisting of C₁₋₁₀-alkyl, phenyl possibly substituted, (CH₂CN₂O)_p, where p ranges from 1 to 4;

Z is chosen in the group consisting of H, N, O, S, SO₃, -CH-, -CH₂- carbon atom, CH₂O, CONH, (CH₂)_qCO₂, where q ranges from 0 to 2;

R₂ is chosen in the group consisting of H, C₁₋₁₂-alkyl, PO(OEt)₂, CH₂CH₂NH₂, aryl, and crown ether, or it forms, with the Z group to which it is bound, a saturated or unsaturated heterocycle, possibly substituted, which may contain up to two hetero-atoms chosen [from] among N, O, and S;

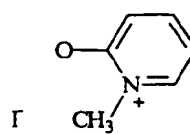
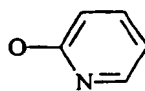
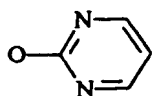
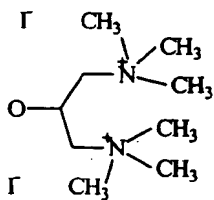
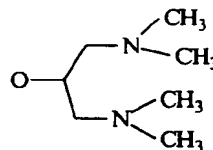
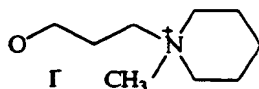
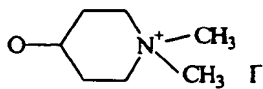
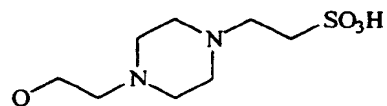
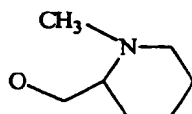
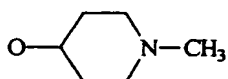
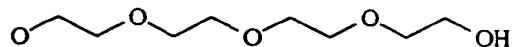
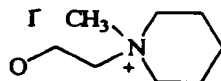
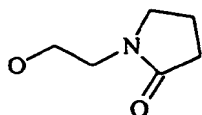
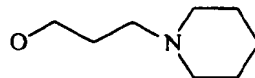
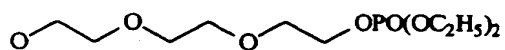
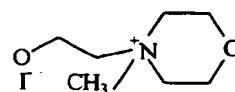
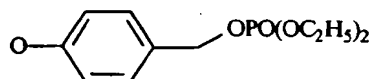
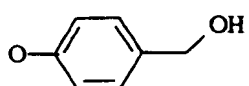
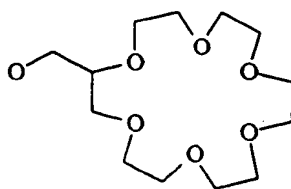
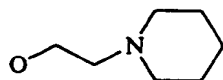
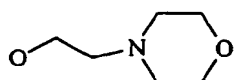
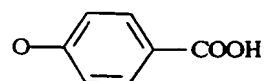
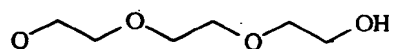
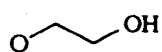
R₃ and R₄, which may be the same or different from one another, are chosen in the group consisting of H, CH₃, and C₂H₅, C₃₋₁₂ alkyl

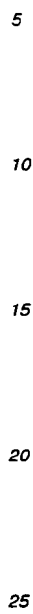
m, n, w, t (independently from one another) are 0 or 1;

v is an integer comprised between 1 and 5;

with the proviso that at least one of R and R₁ is always other than H.

2. Compounds according to Claim 1, in which the saturated or unsaturated heterocycle possibly substituted is chosen in the group consisting of morpholine, piperidine, pyridine, pyrimidine, piperazine, pyrrolidine, and pyrroline.
3. Compounds according to Claim 2, in which the group X-R₅ is chosen in the group consisting of:



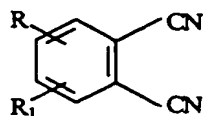


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- (I) a4 in which $R = R_1 = 3\text{-(dimethylamino)phenoxy}$
 (I) a2 in which $R = 3\text{-(dimethylamino)phenoxy}$; $R_1 = H$
 (I) a2 in which $R = 3\text{-(diethylamino)phenoxy}$; $R_1 = H$
 (I) a4 in which $R = R_1 = 3\text{-(diethylamino)phenoxy}$
 (I) a3 in which $R = R_1 = 2\text{-(hydroxy)ethoxy}$
 (I) a1 in which $R = N\text{-(2-aminoethyl)benzamidoyl-4-oxy trifluoro acetate}$; $R_1 = H$
 (I) a1 in which $R = 2\text{-(morpholin-4-yl)ethoxy methylammonium iodide}$; $R_1 = H$
 (I) a1 in which $R = 2\text{-(piperidin-1-yl)ethoxy methylammonium iodide}$; $R_1 = H$
 (I) a2 in which $R = 2\text{-(piperidin-1-yl)ethoxy methylammonium iodide}$; $R_1 = H$
 (I) a2 in which $R = 2\text{-(morpholin-4-yl)ethoxy methylammonium iodide}$; $R_1 = H$
 (I) a3 in which $R = R_1 = 2\text{-(morpholin-4-yl)ethoxy methylammonium iodide}$
 (I) a1 in which $R = 1\text{-methylpiperidinyl-4-oxy methylammonium iodide}$; $R_1 = H$
 (I) a3 in which $R = R_1 = 3\text{-(piperidin-1-yl)propoxy methylammonium iodide}$
 (I) a1 in which $R = 1,3\text{-bis-(dimethylamino)propyl-2-oxy dimethylammonium iodide}$; $R_1 = H$
 (I) a1 in which $R = \text{pyridinyl-2-oxy methylammonium iodide}$; $R_1 = H$
 (I) a1 in which $R = 3\text{-(dimethylamino)phenoxy methylammonium iodide}$; $R_1 = H$
 (I) a1 in which $R = \text{pyridinyl-4-oxy methylammonium iodide}$; $R_1 = H$
 (I) a1 in which $R = (\text{pyridin-4-yl})\text{methoxy methylammonium iodide}$; $R_1 = H$
 (I) a3 in which $R = R_1 = 4\text{-(diethylamino)phenoxy methylammonium iodide}$
 (I) a1 in which $R = 3\text{-(diethylamino)phenoxy methylammonium iodide}$; $R_1 = H$
 (I) a4 in which $R = R_1 = 3\text{-(dimethylamino)phenoxy methylammonium iodide}$
 (I) a4 in which $R = R_1 = 3\text{-(diethylamino)phenoxy methylammonium iodide}$
 (I) a4 in which $R = R_1 = 3\text{-(diethylamino)phenoxy ethylammonium iodide}$
 (I) a2 in which $R = 3\text{-(diethylamino)phenoxy methylammonium iodide}$; $R_1 = H$
 (I) a1 in which $R = 3\text{-(diethylamino)phenoxy ethylammonium iodide}$; $R_1 = H$
 (I) a2 in which $R = 3\text{-(diethylamino)phenoxy ethylammonium iodide}$; $R_1 = H$
 (I) a2 in which $R = 3\text{-(dimethylamino)phenoxy methylammonium iodide}$; $R_1 = H$

6. Compounds of formula (II)



(II)

wherein R and R₁ are as defined in claim 1.

7. Conjugates consisting of a compound of general formula (I) according to Claim 1 and of a macromolecule chosen in the group consisting of polypeptides, proteins and polysaccharides.
8. Process for the preparation of compounds of formula (I) according to Claim 1 wherein the four substituents are identical, in which:
- a polystyrene-based resin is used as substrate for attachment of the phthalodinitrile;
 - the functionalized resin is heated in the presence of Zn (II) salt, with consequent formation of the Zn-phthalocyanine derivative in the solid phase;
 - the Zn-phthalocyanine derivative is cleaved from the resin yielding an impurities free products.
9. Use of a compound according to Claim 1, or of a conjugate according to Claim 7, or of mixtures thereof, possibly in combination with pharmaceutically acceptable excipients, for the preparation of pharmaceutical formulations for the treatment of infectious diseases and diseases characterized by cellular hyperproliferation.
10. Use of according to Claim 9 for the treatment of psoriasis.

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11. Use according to Claim 9 for the treatment of intimal hyperplasia, benign prostate hyperplasia.

12. Use according to Claim 9 for the treatment of atheromas.

13. Use according to Claim 9 for the treatment of viral, fungal or bacterial pathological conditions.

14. Use of a compound according to Claim 1, or of a conjugate according to Claim 7, or of mixtures thereof, possibly in combination with pharmaceutically acceptable excipients, for the preparation of pharmaceutical formulations as diagnostic agents.

15. Pharmaceutical formulation for the treatment of infectious diseases and diseases characterized by cellular hyperproliferation containing a compound according to Claim 1, or a conjugate according to Claim 7 as active principle, possibly in combination with pharmaceutically acceptable excipients.

16. Diagnostic agents containing as active principle a compound according to Claim 1 possibly in combination with a pharmaceutically acceptable carrier.



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EUROPEAN SEARCH REPORT

Application Number
EP 98 11 5036

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The present search report has been drawn up for all claims			
Place of search BERLIN		Date of completion of the search 17 December 1998	Examiner Siatou, E
CATEGORY OF CITED DOCUMENTS X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document			

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EUROPEAN SEARCH REPORT

Application Number
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DOCUMENTS CONSIDERED TO BE RELEVANT			
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Place of search BERLIN		Date of completion of the search 17 December 1998	Examiner Siatou, E
CATEGORY OF CITED DOCUMENTS		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons ----- & : member of the same patent family, corresponding document	
X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document			

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**ANNEX TO THE EUROPEAN SEARCH REPORT
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EP 98 11 5036

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17-12-1998

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